# **On the order and disorder of the transition metal (T) and silicon atoms in ternary thorium transition metal silicides of the**  compositions Th<sub>2</sub>TSi<sub>3</sub> and ThTSi

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(Received October 16, 1993)

#### **Abstract**

During phase investigations of the ternary systems thorium-transition metal-silicon, 12 ternary thorium transition metal silicides Th<sub>2</sub>TSi<sub>3</sub> (T=Mn, Fe, Co, Ni, Cu, Ru, Rh, Pd, Os, Ir, Pt, Au), nine of them for the first time, were prepared by arc melting of the elemental components and subsequent annealing. The silicides with Co, Ni and Cu crystallize with a hexagonal AIB2-type structure, while all other compositions adopt the tetragonal structure of  $\alpha$ -ThSi<sub>2</sub>. Although it can be assumed that the transition metal and silicon atoms occupy the boron and silicon positions of  $AIB_2$  and  $\alpha$ -ThSi<sub>2</sub> with at least some short-range order, no evidence for an ordered occupancy of these positions was found. In contrast, in the new equiatomic compound ThAuSi the gold and silicon atoms occupy the boron positions of  $AIB_2$  in an ordered manner. There is a principal difference for compounds with order-disorder transitions depending on whether the ordering results in a translationengleiche or a klassengleiche subgroup. For structures with a translationengleiche subgroup the order changes the intensities of the reflections already present in the disordered structure, while for ordered structures with ldassengleiche subgroups the order can be detected only through the observation of superstructure reflections.

### **I. Introduction**

Thorium forms two silicides of the composition  $ThSi<sub>2</sub>$ [1-4]: a tetragonal  $\alpha$  form with a structure first determined for this compound and a hexagonal  $\beta$  form which is isotypic with  $\text{AlB}_2$ . In addition, a defect  $\text{AlB}_2$ type structure with a different *c/a* ratio was reported for  $Th_3Si_5$  [2]. Both forms of the disilicide are superconducting at  $T_{cr}$ =3.16 K ( $\alpha$ -ThSi<sub>2</sub>) and  $T_{cr}$ =2.41 K  $(\beta$ -ThSi<sub>2</sub>) respectively [5]. They form solid solutions ThT<sub>r</sub>Si<sub>2-x</sub> (T=Co, Ni, Rh, Ir, Pt) by replacing some percentage of the silicon atoms by transition metal atoms. These five series were intensively investigated

in Bordeaux [6-8] when searching for new superconducting materials.

During the last years a new family of ternary silicides  $U_2TSi$ ,  $T \equiv Mn$ , Fe, Co, Ni, Cu, Ru, Rh, Pd, Os, Ir, Pt, Au) [9-16] was prepared and characterized. These compounds also crystallize either with a hexagonal  $\text{AlB}_2$ type or a tetragonal  $\alpha$ -ThSi<sub>2</sub>-type structure. For U<sub>2</sub>RuSi<sub>3</sub> and  $U_2$ OsSi<sub>3</sub> a new ordered version of an AlB<sub>2</sub>-type structure was observed recently [17].

We were interested in the corresponding thorium compounds and investigated several ternary systems thorium-transition metal-silicon. In the present paper we report on the synthesis of nine new compositions  $Th<sub>2</sub>TSi<sub>3</sub>$  (T = Mn, Fe, Ni, Cu, Ru, Pd, Os, Pt, Au) and on the new equiatomic silicide ThAuSi.

## **2. Experimental details**

Starting materials for the preparation of the ternary silicides were high purity thorium ingots (greater than 99.9%), powders of the transition metals (all with nominal purities greater than 99.9%) and silicon powder (Merck, greater than 99.9%). Thorium filings were prepared under dried paraffin oil. They were washed with dried cyclohexane under argon and were not allowed to contact air prior to the reactions.

The samples were prepared by arc melting of small (about 400 mg) cold-pressed pellets of the elemental components of the intended composition in an argon (99.996%) atmosphere. The argon was further purified by repeatedly melting a titanium button prior to the reactions. The buttons were flipped over and remelted several times to ensure good homogeneity. Subsequently the samples were annealed in evacuated sealed silica tubes at 800 °C for 1 week. The weight loss of the samples after several meltings was always smaller than 1%.

All the crushed buttons had a light grey colour with metallic lustre. They were stable in air and no decomposition was visible after several months.

The samples were characterized by Guinier powder diagrams using Cu  $K\alpha_1$  radiation and  $\alpha$ -quartz  $(a = 491.30 \text{ pm}, c = 540.46 \text{ pm})$  as an internal standard. The lattice constants were obtained by least-squares fits of the powder data.

The powder diffractometer measurement of the ThAuSi sample was performed on a Stoe STADI/P focusing monochromatic beam diffractometer with a rotating sample in the symmetric transmission mode. Cu  $K\alpha_1$  radiation was used with a linear photosensitive detector, a step width of  $0.02^{\circ}$  (2 $\theta$ ) and a constant counting time of 20 s per step. The intensities of this diffraction pattern were obtained with the aid of the RIETAN programme [18].

Density measurements of the ternary samples ThAu<sub>x</sub>Si<sub>2-x</sub> were carried out at room temperature with a conventional pycnometer using petroleum ( $\rho = 0.767$  $g \text{ cm}^{-3}$ ) as the liquid phase.

#### 3. The series with the composition  $Th<sub>2</sub>TSi<sub>3</sub>$

The lattice constants of the 12 silicides with the composition  $Th<sub>2</sub>TSi<sub>3</sub>$  are listed in Table 1. The compounds with Co, Ni and Cu adopt a hexagonal  $\text{AlB}_2$ type structure, while the others crystallize with the tetragonal structure of  $\alpha$ -ThSi<sub>2</sub>. To ensure the proper identification of the phases, intensity calculations for the Guinier powder data were carried out using the positional parameters of the AlB<sub>2</sub>- and  $\alpha$ -ThSi<sub>2</sub>-type structures assuming a statistical distribution of 25% transition metal and 75% silicon atoms on the B and Si positions of  $\text{AlB}_2$  and  $\alpha$ -ThSi<sub>2</sub>. The evaluations of the powder diagrams for  $Th_2MnSi_3$  and  $Th_2OsSi_3$  are shown as examples in Table 2.

We tried to synthesize a silver-containing sample with the composition  $Th_2AgSi_3$ ; however, because of the low boiling point of silver, such samples always showed a large weight loss during the arc-melting process. The powder diagrams of these samples showed the diffraction lines of an  $\text{AlB}_2$ -type cell; however, the lattice constants were close to those of the binary

TABLE 1. Lattice constants of the binary and ternary silicides with hexagonal AlB<sub>2</sub>- and tetragonal  $\alpha$ -ThSi<sub>2</sub>-type structures<sup>a</sup>

Compound	Structure	a	с	c/a	V	Reference
	type	(pm)	(pm)		(nm <sup>3</sup> )	
$\beta$ -ThSi <sub>2</sub>	AIB <sub>2</sub>	413.6(1)	412.6(1)	0.998	0.0611	2
$Th_3Si_5$	AIB <sub>2</sub>	398.5(1)	422.8(1)	1.061	0.0581	2
Th <sub>2</sub> CoSi <sub>3</sub>	AIB <sub>2</sub>	405.2(2)	415.1(1)	1.024	0.0590	This work
$Th_2CoSi_3$	AlB <sub>2</sub>	404.3	418.9	1.036	0.0593	
Th <sub>2</sub> NiSi <sub>2</sub>	AIB <sub>2</sub>	403.22(7)	418.90(6)	1.039	0.0590	This work
Th <sub>2</sub> CuSi <sub>3</sub>	AIB <sub>2</sub>	402.3(1)	419.1(1)	1.042	0.0588	This work
$\alpha$ -ThSi <sub>2</sub>	$\alpha$ -ThSi <sub>2</sub>	412.7	1419.4	3.440	0.2418	
Th <sub>2</sub> MnSi <sub>3</sub>	$\alpha$ ThSi,	410.69(7)	1411.3(3)	3.436	0.2380	This work
Th <sub>2</sub> FeSi <sub>2</sub>	$\alpha$ -ThSi <sub>2</sub>	409.93(8)	1418.5(4)	3.460	0.2384	This work
Th <sub>2</sub> RuSi <sub>3</sub>	$\alpha$ -ThSi,	412.42(5)	1444.7(3)	3.503	0.2457	This work
$Th_2RhSi_3$	$\alpha$ -Thsi <sub>2</sub>	412.41(7)	1438.7(3)	3.489	0.2447	This work
Th <sub>2</sub> RhSi <sub>3</sub>	$\alpha$ -ThSi <sub>2</sub>	411.7	1431	3.476	0.2426	8
$Th_2PdSi_3$	$\alpha$ -ThSi <sub>2</sub>	415.70(5)	1428.2(3)	3.436	0.2468	This work
Th <sub>2</sub> OsSi <sub>3</sub>	$\alpha$ -ThSi <sub>2</sub>	413.84(4)	1437.4(3)	3.473	0.2462	This work
Th <sub>2</sub> IrSi,	$\alpha$ -ThSi <sub>2</sub>	413.66(7)	1436.4(4)	3.472	0.2458	This work
Th <sub>2</sub> IrSi <sub>3</sub>	$\alpha$ -ThSi,	412	1432	3.476	0.2431	8
$Th_2PtSi_3$	$\alpha$ -ThSi <sub>2</sub>	415.92(3)	1428.5(2)	3.435	0.2471	This work
Th <sub>2</sub> AuSi	$\alpha$ -ThSi <sub>2</sub>	419.72(7)	1430.3(4)	3.408	0.2520	This work

aStandard deviations in the positions of the least significant digits are given in parentheses throughout the paper.

TABLE 2. Guinier powder patterns of  $Th<sub>2</sub>MnSi<sub>3</sub>$  and  $Th<sub>2</sub>OsSi<sub>3</sub>$ with a tetragonal  $\alpha$ -ThSi<sub>2</sub>-type structure<sup>a</sup>

Th <sub>2</sub> MnSi <sub>2</sub>				Th <sub>2</sub> OsSi <sub>3</sub>					
hkl	$Q_{c}$	Q.	$I_{\rm c}$	$I_{\alpha}$	hkl	$Q_{\rm c}$	Q.	$I_{\rm c}$	I.
101	643	642	45	S	101	632	632	15	m
004	803	802	25	s	004	774	774	14	s
103	1045	1043	52	s	103	1019	1019	42	s
112	1387	1385	100	VS	112	1361	1359	100	VS
105	1848	1846	42	m	105	1794	1792	47	s
200	2372	2371	32	m	200	2336	2334	37	S
202	2572		$\leq$ 1	÷	202	2529		$\leq$ 1	-
116	2993	2991	12	m	116	2910	2909	13	w
211	3015	3015	13	w	107	2955	2956	24	m
107	3053	3053	21	W	211	2968	<u>.</u>	5	
204	3175	3174	17	m	008	3097		2	-
008	3213		4		204	3110	3109	10	m
213	3416	3418	22	m	213	3355	3353	17	m
206	4179		$\leq 1$		206	4078		${<}1$	~
215	4220	4221	28	m	215	4129	4128	31	m
109	4659	4661	8	vw	109	4504	4505	6	vw

<sup>a</sup>The patterns were recorded in a Guinier camera with Cu  $K\alpha_1$ radiation. The Q values are defined by  $Q = 100/d^2$  (nm<sup>-2</sup>). Intensities were calculated using the LAZY-PULVERIX programme [19] assuming the positional parameters of the  $\alpha$ -ThSi<sub>2</sub> structure with 25% T and 75% Si on the silicon positions. The observed intensities  $I_0$  from very weak to very strong are abbreviated by vw, w, m, s, vs.

compound  $Th_3Si_5[2]$ . Thus we conclude that the samples had a low silver content even though a binary phase ThAg<sub>2</sub> with  $\text{AlB}_2$ -type structure is known to exist [4]. Similar results were obtained when we tried to prepare the corresponding uranium sample  $U_2AgSi_3$ . The lattice constants of these samples were in agreement with those of  $U_3Si_5$  [2].

#### **4. The pseudobinary system ThAu<sub>2</sub>-ThSi<sub>2</sub>**

For the ternary system thorium-gold-silicon the whole pseudobinary section ThAu<sub>2</sub>-ThSi<sub>2</sub> was investigated. The powder diagrams of the samples  $ThAu_xSi_{2-x}$  with x up to 0.7 correspond to the tetragonal  $\alpha$ -ThSi<sub>2</sub>-type structure, while the samples with  $x = 0.9$  and 1.0 show a pattern similar to an  $\widehat{A}$ IB<sub>2</sub>-type structure (Table 3). The samples with a higher gold content contained three phases and thus do not belong to the pseudobinary system. For the samples with the tetragonal  $\alpha$ -ThSi<sub>2</sub>type structure the lattice constant a increases linearly from 412.7 pm  $(\alpha$ -ThSi<sub>2</sub>) to 420.5 pm (ThAu<sub>0.7</sub>Si<sub>1.3</sub>). In contrast, the lattice constant  $c$  first decreases very slightly from 1419.4 pm  $(\alpha$ -ThSi<sub>2</sub>) to 1417.6 pm  $(ThAu<sub>0.2</sub>Si<sub>1.8</sub>)$  and then increases to 1449.5 pm (ThAu<sub>0.7</sub>Si<sub>1.3</sub>). The increase in both lattice constants is certainly due to the larger metallic radius of the gold atoms (144 pm as compared with 132 pm for silicon, both for the coordination number 12 [20]). We also measured the pycnometric densities of these sampies. They agree very well with the calculated ones (Table 3). This excludes the formation of defects as is known for the AlB<sub>2</sub>-type structure of  $Th_3Si_5$  [2]. The densities increase with increasing gold content in a linear fashion (Fig. 1) and this linearity can be extrapolated up to the binary compound  $\text{ThAu}_2$  [4], which also crystallizes in the  $\text{AlB}_2$ -type structure.

The samples with  $x=0.9$  and 1.0 show hexagonal symmetry and the diffraction patterns suggest an  $\text{AlB}_2$ type atomic arrangement. Intensity calculations for the equiatomic composition ThAuSi showed that the gold and silicon atoms occupy the boron positions in this hexagonal cell in an ordered arrangement. The residual values of  $R = 0.026$  for the ordered and  $R = 0.093$  for the random distribution of the gold and silicon atoms on the boron positions of  $\text{AlB}_2$  clearly favour the structure with the ordered arrangement. These residuals, defined by  $R = \sum |I_{o} - I_{c}| / \sum I_{o}$ , were calculated from the intensities listed in Table 4 and do not result from least-squares fits of the data. Thus the structure, as shown in Fig. 2, can be described in the space group *P6m2* (No. 187) with the positional parameters and interatomic distances of Table 5.

#### **5. Discussion**

Of the 12 compositions  $Th<sub>2</sub>TSi<sub>3</sub>$  prepared in the present investigation (Table 1), the nine with  $T = Mn$ ,

TABLE 3. Lattice constants and densities  $\rho$  of the silicides ThAu<sub>x</sub>Si<sub>2-x</sub>

Compound	Structure type	a (pm)	с (pm)	cla	V (nm <sup>3</sup> )	$\rho_{\rm calc}$ $(g \text{ cm}^{-3})$	$\rho_{\rm exp}$ $(g \text{ cm}^{-3})$	Reference
ThS <sub>i</sub>	$\alpha$ -ThSi <sub>2</sub>	412.7	1419.4	3.440	0.2418	7.71	7.63	3, 7
ThAu <sub>0.2</sub> Si <sub>1.8</sub>	$\alpha$ -ThSi <sub>2</sub>	414.65(4)	1417.6(2)	3.419	0.2437	8.78	8.4(1)	This work
ThAu <sub>0.4</sub> Si <sub>1.6</sub>	$\alpha$ -ThSi <sub>2</sub>	417.50(5)	1422.7(2)	3.408	0.2480	9.53	9.2(1)	This work
Th <sub>2</sub> AuSi <sub>2</sub>	$\alpha$ -ThSi <sub>2</sub>	419.71(7)	1430.3(3)	3.408	0.2520	9.82	10.0(1)	This work
ThAu <sub>0.7</sub> Si <sub>1.3</sub>	$\alpha$ -ThSi <sub>2</sub>	420.50(4)	1449.5(2)	3.447	0.2563	10.53	10.5(1)	This work
ThAu <sub>0.9</sub> Si <sub>1.1</sub>	$\mathbf{A}$ i $\mathbf{B}$	425.9(3)	417.3(2)	0.980	0.0655	11.16	11.1(1)	This work
ThAuSi	LiBaSi	426.0(3)	416.4(4)	0.977	0.0655	11.59	11.7(1)	This work
ThAu <sub>2</sub>	$\mathbf{AIB}_2$	474.0(2)	340.2(2)	0.718	0.0662	15.75	15.35	4



Fig. 1. Observed and calculated densities of the solid solutions Th $Au_xSi_{2-x}$ .

TABLE 4. X-Ray powder data for ThAuSi. The pattern was recorded with Cu  $K\alpha_1$  radiation. The last two columns list the calculated intensities assuming ordered and disordered distributions of the Au and Si atoms on the boron sites of  $\text{AlB}_2$ 

hkl	$2\theta_{\rm obs}$	$2\theta_{\rm calc}$	d $(\AA)$	$I_{obs}$	$I_{calc}$ (order)	$I_{\rm calc}$ (disorder)
001	21.35	21.31	4.168	$\mathbf{1}$	1	1
100	24.26	24.22	3.673	242	277	128
101	32.51	32.47	2.756	1000	1000	1000
110	42.65	42.61	2.121	359	354	394
002	43.44	43.40	2.084	113	112	124
111	48.15	48.11	1.890	$\leq 1$	$\leq$ 1	$\leq 1$
200	49.65	49.61	1.837	31	32	27
102	50.35	50.31	1.813	77	80	53
201	54.61	54.57	1.681	218	233	224
112	62.48	62.44	1.486	223	234	232
003	67.39	67.36	1.389	$\leq$ 1	$\leq 1$	1
210	67.45	67.42	1.388	34	34	29
202	68.03	68.00	1.378	26	26	29
211	71.63	71.60	1.317	204	204	191
103	72.76	72.73	1.299	98	98	92
300	78.02	77.99	1.224	63	61	56
301	81.99	81.97	1.175	$\leq$ 1	$\leq 1$	$\leq$ 1
113	83.08	83.06	1.162	$\leq$ 1	$\leq$ 1	${<}1$
212	83.67	83.65	1.155	39	38	40
203	88.14	88.11	1.108	62	62	55
220	93.23	93.21	1.060	41	40	35
302	93.77	93.75	1.056	80	81	69
004	95.38	95.37	1.042	13	13	11
221	97.16	97.14	1.028	$\leq$ 1	$\leq$ 1	9
310	98.29	98.27	1.019	10	10	18
					$R = 0.026$	$R = 0.093$

Fe, Ni, Cu, Ru, Pd, Os, Pt and Au are reported here for the first time. The compounds with Co, Ni and Cu adopt the hexagonal  $\text{AlB}_2$ -type structure, while the others crystallize in the tetragonal structure of  $\alpha$ -ThSi<sub>2</sub>. This is in contrast with the corresponding uranium compounds, where only  $U_2CuSi_3$  [13-15] adopts the



Fig. 2. Crystal structure of ThAuSi projected along the z direction. All atoms are situated on mirror planes at  $z=0$  and  $\frac{1}{2}$  connected by thin and thick lines respectively.

TABLE 5. Positional parameters and interatomic distances (picometres) of ThAuSi

Atom	$P\bar{6}m2$		x	y	z	
Th	1a		0	o	O	
Au	1f		$rac{2}{3}$	$\frac{1}{3}$	$\frac{1}{2}$	
1d Si			$\frac{1}{3}$		$\frac{1}{2}$	
Th-6 Au 6 Si 2 Th 6 Th	322.3(3) 322.3(3) 416.4(4) 426.0(3)	Au-3 Si 6 Th	246.0(2) 322.3(3)	$Si-3$ Au 6 Th	246.0(2) 322.3(3)	

tetragonal  $\alpha$ -ThSi<sub>2</sub>-type structure, while the others crystallize in the hexagonal structure of  $\text{AlB}_2$ . Considerable homogeneity ranges may be assumed for these phases and those with the tetragonal  $\alpha$ -ThSi<sub>2</sub>-type structure may form continuous series of solid solutions with the binary phase  $\alpha$ -ThSi<sub>2</sub>, as was shown already for the series ThRh<sub>x</sub>Si<sub>2-x</sub> and ThIr<sub>x</sub>Si<sub>2-x</sub> [6, 8]. At temperatures above 1350 °C  $\alpha$ -ThSi<sub>2</sub> transforms to the  $\beta$  form [4] and similar transformations with temperature and/or pressure may be assumed to occur also for the ternary compositions  $Th<sub>2</sub>TSi<sub>3</sub>$  reported here.

Of the pseudobinary systems  $\alpha$ -ThSi<sub>2</sub>-ThT<sub>2</sub>, most information is available for the systems with  $T = Co$ , Ni, Rh, Ir, Pt and Au. In Fig. 3 we give a summary of the structures occurring in these pseudobinary systems. It can be seen that a change in structure occurs quite frequently and the system with nickel even has two phases with  $\alpha$ -ThSi<sub>2</sub>- and two with AlB<sub>2</sub>-type structure at relatively low temperatures (the samples of  $Th<sub>2</sub>NiSi<sub>3</sub>$  and ThNiSi were annealed at 800 and 950 °C respectively). At higher temperatures other equilibria may exist. In fact, for the systems with nickel and gold it seems possible that continuous solid solutions with  $AlB_2$ -type structure might form at very high temperatures, since above 1350 °C ThSi<sub>2</sub> has the hexagonal  $\text{AlB}_2$ -type structure like ThNi<sub>2</sub> and ThAu<sub>2</sub>. Furthermore,



Thlr 2 has a MgCu2-type **structure** 

Fig. 3. Crystal structures occurring for certain compositions of the pseudobinary sections  $\alpha$ -ThSi<sub>2</sub>-ThT<sub>2</sub>.

it is frequently observed that at higher temperatures the structures with the smaller cell volumes and higher symmetry (smaller number of positional parameters) are favoured, *i.e.* the  $\text{AlB}_2$ -type structure is favoured for the ternary compositions also for these reasons.

For ThAuSi we have established a structure deriving from that of  $AIB<sub>2</sub>$  with an ordered arrangement of the Au and Si atoms on the boron positions of  $\text{AlB}_2$ . Thus the Au and Si atoms alternate within the graphite-like nets, as is known for the hexagonal modification of BN. The atomic positions of ThAuSi, as we described them in Table 5, correspond to those of LiBaSi [21] and SrPtSb [22], where Li, Si, Pt and Sb atoms occupy the boron positions of  $\text{AlB}_2$  in an ordered manner.

A similar structure with a doubled c axis was reported recently for ScAuSi and LuAuSi [23]. In these silicides the doubling of the  $c$  axis is solely a result of a slight puckering of the BN-type AuSi layers. Such puckered nets also occur in the LiGaGe-type structure [24] of YAuSi [23]. In this latter structure the doubling of the c axis additionally results from the alternating arrangement of the Au and Si atoms when viewed along the

c axis. While we can rule out an LiGaGe-type structure for ThAuSi right away  $-$  because in the AlB<sub>2</sub>-type subcell of such a structure the Au and Si atoms again would be statistically distributed  $-$  we cannot exclude a slight puckering of the AuSi nets of the type found for ScAuSi and LuAuSi in ThAuSi. The superstructure reflections (resulting from the doubled  $c$  axis) for such a structure would be very weak and hardly visible in a powder pattern. In fact, no such superstructure lines could be detected in the pattern of ThAuSi. However, a comparison of the Au-Si distances at first sight slightly favours a puckered arrangement also for ThAuSi. In the puckered AuSi nets of ScAuSi, YAuSi and LuAuSi these distances are  $249(1)$ ,  $249(1)$  and  $252(1)$  pm respectively. Thus they are all slightly greater than the distance of 246.0(3) pm calculated from the positional parameters of ThAuSi as given in Table 5, and longer Au-Si distances would result if the AuSi nets were puckered. However, the Au-Si bond distances of the four structures do not need to be the same, because Th certainly will contribute four valence electrons per formula unit to the band structure, while Sc, Y and Lu can only contribute three. In the absence of any hard evidence for the puckering of the AuSi nets in ThAuSi we therefore favour the simpler description of the structure as given in Table 5.

For the equiatomic silicide compositions ThTSi of Fig. 3 with a tetragonal  $\alpha$ -ThSi<sub>2</sub>-type structure an ordered arrangement of the T and Si atoms on the silicon positions of  $\alpha$ -ThSi<sub>2</sub> has been established so far only for ThPtSi [7], which was found to be isotypic with LaPtSi [25]. No conclusive evidence for such an atomic order was found for the compositions with  $T = Co$ , Rh, Ir and Ni; nevertheless, it was not ruled out [7]. Actually, in view of the presently available evidence, we expect at least short-range order for all these compositions.

For the silicides with the composition 2:1:3 two closely related superstructures were found, both with  $\text{AlB}_2$ type subcells. One of these was reported for  $Er<sub>2</sub>RhSi<sub>3</sub>$ [26, 27] and the rare earth (RE) transition metal silicides with the compositions  $RE_2RhSi_3$  [26] and  $RE_2PdSi_3$ [28] are believed to be isotypic with  $Er<sub>2</sub>RhSi<sub>3</sub>$ . In this structure both the *a* and the *c* axis of the  $AlB_2$ -type subcell are doubled. For the other superstructure, investigated only recently for  $U_2RuSi_3$  [17], only the a axis of the  $AIB_2$ -type subcell was established to be doubled, whereas no evidence for the doubling of the c axis could be detected. Both structures have similar atomic positions with  $TSi<sub>3</sub>$  nets as shown in the righthand part of Fig. 4. The doubling of the  $c$  axis in  $Er<sub>2</sub>RhSi<sub>3</sub>$  results from slight distortions of the  $U<sub>2</sub>RuSi<sub>3</sub>$ structure and possibly both of these structures occur for several of the 2:1:3 silicides, the higher symmetry  $U_2RuSi_3$ -type structure at high temperatures and



Fig. 4. Two-dimensionally infinite transition metal-silicon networks  $(AuSi)_n$  and  $(TSi_3)_n$  in the structures of ThAuSi and Er<sub>2</sub>RhSi<sub>3</sub> [26, 27] as well as  $U_2RuSi_3$  [17]. While the gold atoms are on the B positions of ThAuSi with ordered AlB<sub>2</sub> (LiBaSi)-type structure, the rhodium and ruthenium atoms of  $Er<sub>2</sub>RhSi<sub>3</sub>$  and  $U_2RuSi_3$  occupy the B and C positions of the AlB<sub>2</sub>-type subcells (dashed lines) in equal amounts.

the lower symmetry  $Er<sub>2</sub>RhSi<sub>3</sub>$ -type structure at low temperatures.

In the present investigation we have found the samples  $Th<sub>2</sub>TSi<sub>3</sub>$  (T=Co, Ni, Cu) to crystallize with an AlB<sub>2</sub>type structure. We had expected some order for the transition metal and silicon atoms on the boron sites of the  $\text{AlB}_2$ -type cell. However, no evidence for a superstructure could be detected even though the samples were annealed for 1 week at 800 °C. We have calculated powder patterns assuming ordered arrangements of the types found for  $Er<sub>2</sub>RhSi<sub>3</sub>$  and  $U<sub>2</sub>RuSi<sub>3</sub>$ , and these show that the scattering powers of the transition metal atoms Co, Ni and Cu are too similar to that of silicon to produce well-visible superstructure reflections in the presence of the strongly scattering thorium atoms. Also, one has to keep in mind that sharp superstructure reflections can develop only for samples with exactly the composition 2:1:3. Slight deviations from this ideal composition may prevent longrange order and thus broaden the superstructure reflections, which makes them even less visible.

This is different for the atomic order of the LiBaSi type found for ThAuSi, where the order expresses itself already in the intensities of the  $AIB_2$ -type subcell reflections. For an atomic order of the type found for  $Er<sub>2</sub>RhSi<sub>3</sub>$  and  $U<sub>2</sub>RuSi<sub>3</sub>$  the intensities of the AlB<sub>2</sub>-type subcell remain unchanged, since the transition metal and silicon atoms are occupying the B and C positions (Fig. 4) of the  $\text{AlB}_2$ -type subcell in equal amounts. Thus there is a principal difference between the two cases. For both cases the ordering of the transition metal and silicon atoms on the boron positions of the AIB2-type subcell (space group *P6/mmm,* crystal class *6/mmm)* results in a reduction of the symmetry. In the case of the LiBaSi-type ordered structure (space group *P6m2,* crystal class 6m2) the translational symmetry does not change. The ordering results in a *"transla-* *tionengleiche"* subgroup [29] and expresses itself already in the intensities of the subcell reflections. In the other case of the  $Er<sub>2</sub>RhSi<sub>3</sub>$ - (space group  $P6<sub>3</sub>/mmc$ , crystal class 6/mmm) and U<sub>2</sub>RuSi<sub>3</sub>-type (space group *P6/mmm*, crystal class *6/mmm)* superstructures the translational symmetry is changed (hence the name "super" structure) and the crystal class has not changed. The ordering results in a *"klassengleiche"* subgroup and can be detected only through the occurrence of superstructure reflections. To our knowledge this principal difference in (potential) order-disorder transitions has not been emphasized before. In the classical cases of or $der-disorder$  transitions  $(Cu<sub>3</sub>Au, CuZn)$  the ordering results always in *"klassengleiche"* subgroups and superstructure reflections.

For the other compositions 2:1:3 with the tetragonal  $\alpha$ -ThSi<sub>2</sub>-type structure we also could not find any superstructure reflections, although at least some shortrange order can be expected. For this subcell a superstructure was established up to now only for the composition 1:1:1 with the LaPtSi-type structure [25], as already mentioned. We think that it might be worthwhile to investigate the whole pseudobinary sections of these ternary systems also at different temperatures and pressures to obtain more information about the relative stabilities of the AlB<sub>2</sub>- and  $\alpha$ -ThSi<sub>2</sub>-type structures and the ordering of the transition metal and silicon atoms.

#### **Acknowledgments**

We thank K. Wagner for the collection of the powder diffractometer data of ThAuSi. We also acknowledge Dr. W. Gerhartz (Degussa AG.) and Dr. G. Höfer (Heraeus Quarzschmelze) for generous gifts of platinum metals and silica tubes. We are grateful to the Fonds der Chemischen Industrie for a stipend to J.H.A. and to the European Community for a fellowship to R.P. in the framework of the "Human Capital and Mobility Programme". This work was supported by the Deutsche Forschungsgemeinschaft.

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